SCIENCE - Chemistry

Key Stage	Year	Autumn 1	Autumn 2	Spring 1	Spring 2	Summer 1	Summer 2
KS3	7	Lab Safety How to use a Bunsen Burner Apprentice Scientists - Scientific Skills Variables Collecting data Analysis of data	Atoms, Elements and Compounds Solids, Liquids and gases Atoms and elements Compounds The Periodic Table	Chemical Reactions Thermal Decomposition Combustion Oxidation Acids and Alkalis Neutralisation	The Earth's Atmosphere The structure of the earth The water cycle The Carbon cycle Testing for gases Recycling		
	8	Introductory lessons - scientific skills Data collection - range and repeats Scientific communication Evidence and theories Risks and benefits The Earth's Structure Igneous rocks Sedimentary rock Metamorphic rock The rock cycle Fossil fuels		Mixtures • Pure Substances • Separating Mixtures • Chromatography • Dissolving		Metals and Acids Symbols Equations The Reactivity of metals The Extraction of metals Metals and acids Bases and acids Making a soluble salt	
	9	Scientific skills - Tables Graphs Calculating rates	Exo or Endo Reversible reactions Exothermic or endothermic Bond energy calculations	Materials	GCSE begins CH1 (Spec 4.1) Atomic structure and the periodic table		table

	After each half term, students will have a short assessment.										
KS4	10	CH1 (Spec 4.1) Atomic structure and the periodic table CH2 (Spec 4.2) Bonding, Structure and the Properties of Matter	CH2 (Spec 4.2) Bonding, Structure and the Properties of Matter CH3 (Spec 4.3) Quantitative Chemistry	CH3 (Spec 4.3) Quantitative Chemistry CH4 (Spec 4.4) Chemical Changes	CH4 (Spec 4.4) Chemical Changes	CH5 (Spec 4.5) Energy Changes	EXAMS CH6 (Spec 4.6) The Rate and Extent of Chemical Change				
	11	CH6 (Spec 4.6) The Rate and Extent of Chemical Change CH7 (Spec 4.7) Organic Chemistry	EXAMS CH7 (Spec 4.7) Organic Chemistry	CH8 (Spec 4.8) Chemical analysis CH9 (Spec 4.9) The Earth's Atmosphere CH10 (Spec 4.10) Using Our Resources	CH10 (Spec 4.10) Using Our Resources REVISION	REVISION					
		After e	each topic, students w	vill have an assessme	nt - ranging betwee	n 30-45 marks.	•				
	12	Foundations in Chemistry - atomic structure, quantitative chemistry, electrons and bonding.	Foundations in Chemistry Periodic Table and Energy Core Organic Chemistry	Periodic Table and Energy Core Organic Chemistry	Periodic Table and Energy Core Organic Chemistry	Periodic Table and Energy Core Organic Chemistry	EXAMS Physical chemistry and transition elements Organic chemistry and analysis				
KS5	13	Physical chemistry and transition elements (Redox and TM complex ions) Organic chemistry and analysis (benzene and carbonyls)	EXAMS Physical chemistry and transition elements (Rates and Equilibrium) Organic chemistry and analysis (amines, amides and amino acids)	Physical chemistry and transition elements (acids and buffers, entropy) Organic chemistry and analysis (Synthesis and analysis)	EXAMS Content finalised - REVISION						